

the remaining atoms as incorporated in *SHELX76* (Sheldrick, 1976). All calculations on SUN4/280 computer system. Atomic parameters are given in Table 1, selected interatomic parameters in Table 2* and numbering scheme used is shown in Fig. 1 [drawn with *ORTEP* (Johnson, 1971) at 15% probability levels].

Related literature. Dimeric ring structures similar to that observed in [Ag{Ph₂P(CH₂)₃PPh₂}NO₃]₂ are well known in Ag chemistry for compounds of the general formula [Ag(P—P)X]₂ where X is a uninegative anion. Eight-membered rings have been characterized crystallographically in two bis(diphenylphosphino)methane (dppm) complexes, *i.e.* [Ag(dppm)NO₃]₂ (Ho & Bau, 1983; Tiekkink, 1990) and in the bis(dimethylphosphino)methane analogue [Ag(dmpm)NO₃]₂ (Karsch & Schubert, 1982).

The Australian Research Council is thanked for support.

* Lists of structure factors, anisotropic thermal parameters, non-H atom bond lengths and angles and H-atom coordinates have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 53051 (15 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

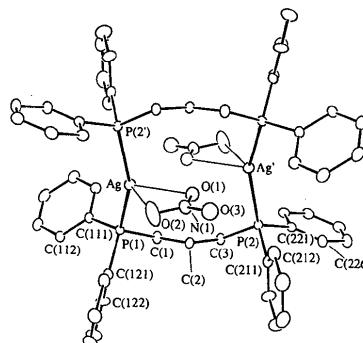


Fig. 1. Molecular structure and crystallographic numbering scheme employed for [Ag{Ph₂P(CH₂)₃PPh₂}NO₃]₂ (Johnson, 1971).

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Structure of a Nucleoside Analogue, 3'-Deoxy-2'-fluorothymidine*

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Abstract. 1-(2,3-Dideoxy-2-fluoro-β-D-erythro-pentofuranosyl)thymine, C₁₀H₁₃FN₂O₄, $M_r = 244.22$, monoclinic, $P2_1$, $a = 10.6001 (7)$, $b = 6.1847 (1)$, $c = 16.9227 (4)$ Å, $\beta = 98.040 (3)$ °, $V = 1098.52 (8)$ Å³, $Z = 4$, $D_m = 1.47$, $D_x = 1.477$ Mg m⁻³, Ni-filtered

Cu $K\alpha$ radiation, $\lambda = 1.54178$ Å, $\mu = 1.040$ mm⁻¹, $F(000) = 512$, $T = 298$ K, final $R = 0.033$ for 1868 unique observed reflections. The asymmetric unit contains two molecules (*A* and *B*). For molecule *A*: the *N*-glycosidic torsion angle χ has a value of -155.2 (2)° in the *anti* range; the sugar pucker is ³T₂ with $P = 12 (1)$ ° and $\psi_m = 35 (1)$ ° and the C4'—C5' conformation is +sc with $\gamma = 55.1 (3)$ °. For molecule *B*: the *N*-glycosidic torsion angle χ has a value

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of $-168.0(2)^\circ$ in the *anti* range; the sugar pucker is 3T_2 with $P = 8(1)^\circ$ and $\psi_m = 35(1)^\circ$ and the C4'—C5' conformation is +sc with $\gamma = 49.7(3)^\circ$. The conformational parameters are in accordance with the IUPAC-IUB Joint Commission on Biochemical Nomenclature [*Pure Appl. Chem.* (1983), **55**, 1273–1280] guidelines. There is little difference in sugar puckering (P and ψ_m) between molecules *A* and *B*. The main difference lies in the *N*-glycosidic torsion angle χ .

Experimental. Colourless prismatic crystals obtained at room temperature from an acetone–hexane solution, dimensions $\sim 0.5 \times 0.3 \times 0.2$ mm. Density measured by flotation in *n*-heptane/CCl₄. Hilger & Watts computer-controlled four-circle diffractometer, Ni-filtered Cu $K\alpha$ radiation, $\omega/2\theta$ scan technique ($2\theta_{\max} = 130^\circ$, $-12 \leq h \leq 12$, $-7 \leq k \leq 0$, $-20 \leq l \leq 20$). Cell dimensions by least-squares refinement of the setting angles of 24 reflections with $19 < 2\theta < 50^\circ$, space group $P2_1$ from systematic absences $0k0$ for k odd. Four standard reflections (422, 125, 120, 315) monitored after every 50 reflections showed no significant decrease in intensity per hour, 3997 reflections measured, 2011 unique reflections ($R_{\text{int}} = 0.007$) of which 1872 were considered observed [$F > 6\sigma(F)$]. Four reflections (004, 011, 200, 211) badly affected by extinction were eliminated. Lorentz–polarization corrections, absorption corrections by the method of North, Phillips & Mathews (1968) based on one reflection (040) with values between 0.9965 and 0.8838, scattering factors from *International Tables for X-ray Crystallography* (1974) and Stewart, Davidson & Simpson (1965) (for H). The structure was solved by *MULTAN82* (Main, Fiske, Hull, Lessinger, Germain, Declercq & Woolfson, 1982). The *E* map calculated from the solution with the best figure of merit revealed 17 of the 34 non-H atoms in the asymmetric unit. The remaining atoms were obtained from a subsequent Fourier synthesis. A difference synthesis revealed the

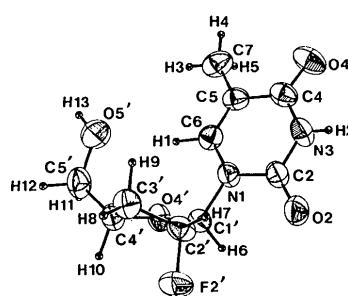


Fig. 1. A view of the title compound with 50% probability anisotropic displacement ellipsoids for the non-H atoms and atomic numbering scheme.

Table 1. *Atomic coordinates ($\times 10^4$) and equivalent isotropic thermal parameters with e.s.d.'s of the refined parameters in parentheses*

	x	y	z	$B_{\text{eq}}(\text{\AA}^2)$
N1 <i>A</i>	9228 (2)	6140*	2627 (1)	3.24 (4)
C2 <i>A</i>	9842 (2)	7627 (4)	3142 (1)	3.20 (4)
O2 <i>A</i>	10432 (2)	9161 (3)	2932 (1)	4.27 (4)
N3 <i>A</i>	9750 (2)	7228 (4)	3934 (1)	3.51 (4)
C4 <i>A</i>	9141 (2)	5519 (4)	4237 (1)	3.35 (4)
O4 <i>A</i>	9164 (2)	5376 (4)	4973.2 (9)	4.54 (4)
C5 <i>A</i>	8510 (2)	4032 (5)	3662 (1)	3.38 (5)
C6 <i>A</i>	8587 (2)	4419 (5)	2887 (1)	3.43 (5)
C7 <i>A</i>	7799 (3)	2133 (5)	3923 (2)	4.60 (6)
C1 <i>A'</i>	9350 (3)	6422 (5)	1763 (1)	3.77 (5)
C2 <i>A'</i>	10526 (2)	5293 (6)	1565 (1)	4.59 (6)
F2 <i>A'</i>	10953 (2)	6435 (4)	932 (1)	7.73 (5)
C3 <i>A'</i>	10040 (2)	3136 (6)	1259 (2)	4.42 (6)
C4 <i>A'</i>	8730 (2)	3674 (5)	840 (1)	3.72 (5)
O4 <i>A'</i>	8309 (2)	5459 (3)	1292.8 (9)	3.85 (3)
C5 <i>A'</i>	7755 (3)	1887 (6)	775 (2)	4.81 (6)
O5 <i>A'</i>	7664 (2)	1082 (4)	1542 (1)	6.43 (5)
N1 <i>B</i>	4865 (2)	5186 (4)	2405 (1)	3.50 (4)
C2 <i>B</i>	5433 (2)	6399 (5)	1866 (1)	3.68 (5)
O2 <i>B</i>	6084 (2)	7974 (4)	2076 (1)	4.73 (4)
N3 <i>B</i>	5186 (2)	5722 (4)	1095 (1)	4.19 (4)
C4 <i>B</i>	4472 (2)	3973 (5)	815 (1)	4.09 (5)
O4 <i>B</i>	4316 (2)	3588 (4)	87 (1)	5.79 (5)
C5 <i>B</i>	3938 (2)	2721 (5)	1401 (2)	3.94 (5)
C6 <i>B</i>	4150 (2)	3383 (5)	2170 (2)	3.66 (5)
C7 <i>B</i>	3198 (3)	734 (6)	1141 (2)	5.68 (7)
C1 <i>B'</i>	5134 (2)	5858 (5)	3255 (1)	3.40 (5)
C2 <i>B'</i>	4424 (3)	7897 (4)	3407 (1)	3.79 (5)
F2 <i>B'</i>	5151 (2)	8951 (3)	4049 (1)	5.93 (4)
C3 <i>B'</i>	3225 (3)	7057 (5)	3670 (2)	4.05 (5)
C4 <i>B'</i>	3667 (2)	5051 (5)	4130 (1)	3.74 (5)
O4 <i>B'</i>	4716 (2)	4239 (3)	3735.5 (9)	3.75 (3)
C5 <i>B'</i>	2694 (3)	3299 (6)	4155 (2)	4.93 (6)
O5 <i>B'</i>	2074 (2)	2840 (4)	3377 (1)	6.04 (5)

* Parameter kept fixed for origin definition.

Table 2. *Bond lengths (\AA), bond angles ($^\circ$) and selected torsion angles ($^\circ$) with e.s.d.'s in parentheses*

	<i>A</i>	<i>B</i>	<i>A</i>	<i>B</i>
N1—C2	1.366 (3)	1.381 (3)	C5—C7	1.495 (4)
N1—C6	1.370 (3)	1.376 (3)	C1'—C2'	1.507 (4)
N1—C1'	1.496 (3)	1.486 (3)	C1'—O4'	1.400 (3)
C2—O2	1.217 (3)	1.217 (3)	C2'—F2'	1.409 (3)
C2—N3	1.380 (3)	1.360 (4)	C3'—C3'	1.496 (5)
N3—C4	1.374 (3)	1.367 (4)	C3'—C4'	1.506 (3)
C4—O4	1.246 (3)	1.243 (4)	C4'—O4'	1.450 (3)
C4—C5	1.434 (3)	1.437 (4)	C4'—C5'	1.506 (4)
C5—C6	1.347 (3)	1.352 (4)	C5'—O5'	1.406 (4)
	<i>A</i>	<i>B</i>	<i>A</i>	<i>B</i>
C2—N1—C6	122.2 (2)	121.6 (2)	N1—C6—C5	123.4 (2)
C2—N1—Cl'	116.6 (2)	116.4 (3)	N1—C1'—C2'	110.5 (2)
C6—N1—Cl'	121.3 (2)	122.0 (3)	N1—C1'—O4'	109.7 (2)
N1—C2—O2	123.8 (3)	121.6 (2)	C2—C1'—O4'	106.7 (3)
N1—C2—N3	114.0 (2)	115.0 (2)	C1'—C2'—F2'	107.5 (2)
O2—C2—N3	122.3 (2)	123.4 (2)	C1'—C2'—C3'	103.6 (2)
C2—N3—C4	126.9 (3)	126.9 (3)	F2'—C2'—C3'	108.4 (2)
N3—C4—O4	118.9 (2)	119.1 (3)	C2'—C3'—C4'	102.5 (2)
N3—C4—C5	116.1 (2)	116.1 (2)	C3'—C4'—O4'	104.5 (2)
O4—C4—C5	125.1 (2)	124.9 (3)	C3'—C4'—C5'	116.8 (3)
C4—C5—C6	117.4 (2)	118.2 (2)	O4'—C4'—C5'	109.8 (2)
C4—C5—C7	120.6 (2)	118.8 (2)	C1'—O4'—C4'	110.4 (2)
C6—C5—C7	121.9 (2)	122.9 (3)	C4'—C5'—O5'	109.1 (2)
	<i>A</i>	<i>B</i>	<i>A</i>	<i>B</i>
C2—N1—C1'—O4'	—155.24 (0.19)	—168.02 (0.20)		
C2'—C1'—O4'—C4'	3.79 (0.26)	6.14 (0.25)		
O4'—C1'—C2'—C3'	—24.14 (0.27)	—26.07 (0.25)		
C1'—C2'—C3'—C4'	34.07 (0.27)	35.07 (0.25)		
C2'—C3'—C4'—O4'	—32.14 (0.26)	—31.81 (0.24)		
C3'—C4'—O4'—C1'	18.01 (0.26)	16.22 (0.25)		
C3'—C4'—C5'—O5'	55.07 (0.33)	49.70 (0.33)		

	<i>A</i>	<i>B</i>
C2—N1—C1'—O4'	—155.24 (0.19)	—168.02 (0.20)
C2'—C1'—O4'—C4'	3.79 (0.26)	6.14 (0.25)
O4'—C1'—C2'—C3'	—24.14 (0.27)	—26.07 (0.25)
C1'—C2'—C3'—C4'	34.07 (0.27)	35.07 (0.25)
C2'—C3'—C4'—O4'	—32.14 (0.26)	—31.81 (0.24)
C3'—C4'—O4'—C1'	18.01 (0.26)	16.22 (0.25)
C3'—C4'—C5'—O5'	55.07 (0.33)	49.70 (0.33)

position of two H atoms (attached to atoms O^{5'A} and O^{5'B}). All the other H atoms were included at calculated positions (C—H and N—H distance = 0.95 Å). All H atoms were refined with fixed isotropic temperature factors ($B_{eq} = 4 \text{ \AA}^2$). All other atoms were refined anisotropically on F by full-matrix least squares. The refinement converged at $R = 0.033$, $wR = 0.041$, $S = 0.264$, $w = (C_0 + C_1|F_o| + C_2|F_o|^2 + C_3|F_o|^3)^{-1}$, where $C_0 = 1000$, $C_1 = 1$, $C_2 = 0.001$, $C_3 = 0.000015$. 385 refined parameters, maximum shift/e.s.d. = 0.03, minimum and maximum electron density -0.183 and 0.129 e Å⁻³. The number of reflections per refined variable was 1868/385 = 4.9. All calculations were performed on a PDP 11/73 microcomputer using SDP/PDP (Enraf-Nonius, 1982) and PARST (Nardelli, 1983). An ORTEP view (Johnson, 1976) of the title compound with the atomic numbering scheme is shown in Fig. 1.* The final fractional atomic coordinates are given in Table 1. Bond lengths, bond angles and some selected torsion angles are given in Table 2.

* Lists of structure factors, anisotropic thermal parameters, bond lengths and angles involving H atoms, least-squares planes, and H-atom parameters have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 52902 (21 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

Related literature. The method of preparation of this possible anti-AIDS compound has been described by Van Aerschot, Herdewijn, Balzarini, Pauwels & De Clercq (1989).

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Structure of a Nucleoside Analogue, 2',3'-Dideoxy-3'-fluoro-5-iodouridine*

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Abstract. 1-(2,3-Dideoxy-3-fluoro-β-D-erythro-pentofuranosyl)-5-iodouracil, C₉H₁₀FIN₂O₄, $M_r = 356.09$, orthorhombic, $B22_12$, $a = 7.917(4)$, $b = 16.962(8)$, $c = 17.694(9)$ Å, $V = 2377(2)$ Å³, $Z = 8$, $D_m = 1.98$, $D_x = 1.990$ Mg m⁻³, graphite-monochro-

* Structural Studies of Modified Nucleosides. Part IV. Part III: Everaert, Peeters, Blaton, De Ranter, Van Aerschot & Herdewijn (1990).

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mated Mo Kα radiation, $\lambda = 0.71073$ Å, $\mu = 2.680$ mm⁻¹, $F(000) = 1376$, $T = 290$ K, final $R = 0.022$ for 1106 unique observed reflections. The N-glycosidic torsion angle χ has a value of -107.0 (4)° in the *anti* range; the sugar pucker is $\ddot{\gamma}T$ with $P = 184(1)$ ° and $\psi_m = 29(1)$ ° and the C4'—C5' conformation is +sc with $\gamma = 43.8(6)$ °. The conformational parameters are in accordance with the IUPAC-IUB Joint Commission on Biochemical